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FIRST NAMED INVENTOR APPLICATION NO. FILING DATE ATTORNEY DOCKET NO. CONFIRMATION NO. 10/049,705 06/17/2002 A-71327/DJB/MAK 7594 Karl Foger EXAMINER 12/12/2003 Flehr Hohbach Test ALEJANDRO, RAYMOND Albritton & Herbert ART UNIT PAPER NUMBER **Suite 3400** Four Embarcadero Center 1745 San Francisco, CA 94111-4187 DATE MAILED: 12/12/2003

Please find below and/or attached an Office communication concerning this application or proceeding.

•		—		Applicati n N .	Applicant(s)	,
•'	Offic	Action Summary		10/049,705	FOGER ET AL.	
				Examiner	Art Unit	
				Raymond Alejandro	1745	····
Period f	The MAII or Reply	LING DATE of this comm	unicati n app	ears on the cover shee	t with the correspondence addre	ss
THE - Extraording - If th - If N - Fail - Any	MAILING I ensions of time r or SIX (6) MONT the period for repl O period for repl ture to reply with the reply received by	y is specified above, the maximum	INICATION. ons of 37 CFR 1.13 mmunication. y (30) days, a reply n statutory period w eply will, by statute, ns after the mailing	i6(a). In no event, however, ma within the statutory minimum of ill apply and will expire SIX (6) I cause the application to becom	y a reply be timely filed fithirty (30) days will be considered timely. MONTHS from the mailing date of this committee of the committee of t	unication.
1)⊠	Responsi	ve to communication(s)	filed on <u>17 Ju</u>	ne 2002.		
2a) <u></u>	This actio	n is FINAL .	2b)⊠ This a	action is non-final.		
3)□		application is in condition			natters, prosecution as to the mo C.D. 11, 453 O.G. 213.	erits is
Disposit	tion of Clai	ms				
4)🖂	Claim(s) 1	<u>-21</u> is/are pending in the	e application.			
	4a) Of the	above claim(s) is	/are withdraw	n from consideration.	•	
5)[) Claim(s) is/are allowed.					
6)⊠	Claim(s) 1	/-21 is/are rejected.				
7)	Claim(s)_	is/are objected to.				
8)[Claim(s) _	are subject to rest	triction and/or	election requirement.		
Applicat	tion Papers	;				
9)🖂	The specif	ication is objected to by	the Examiner	•		
10)⊠	The drawir	ng(s) filed on <u>17 June 20</u>	002 is/are: a)	☐ accepted or b)⊠ o	bjected to by the Examiner.	
	Applicant n	nay not request that any ob	jection to the o	lrawing(s) be held in abe	yance. See 37 CFR 1.85(a).	
	Replaceme	ent drawing sheet(s) includi	ing the correcti	on is required if the draw	ing(s) is objected to. See 37 CFR 1	.121(d).
11)	The oath o	r declaration is objected	I to by the Exa	aminer. Note the attac	hed Office Action or form PTO-	152.
Priority	under 35 U	.S.C. §§ 119 and 120				
12)⊠	Acknowle	dgment is made of a cla]Some * c)☐ None of		priority under 35 U.S.	C. § 119(a)-(d) or (f).	
u,		tified copies of the priori		have been received.		
* (2.☐ Cer 3.☒ Cop app	tified copies of the priori pies of the certified copie lication from the Interna	ty documents s of the priori tional Bureau	have been received in ity documents have be (PCT Rule 17.2(a)).	n Application No en received in this National Sta	ge
13) <u> </u>	Acknowledg since a spec 37 CFR 1.78	cific reference was includ 3.	n for domestic ded in the firs	priority under 35 U.S. t sentence of the spec	C. § 119(e) (to a provisional ap ification or in an Application Dat	
		anslation of the foreign I				
					C. §§ 120 and/or 121 since a spaper Application Data Sheet. 37 CFF	
Attachmer	nt(s)					
2) Notic	ce of Draftspe	es Cited (PTO-892) rson's Patent Drawing Review sure Statement(s) (PTO-1449)	(PTO-948)) Paper No(s) <u>7</u> .	5) Notice	w Summary (PTO-413) Paper No(s) of Informal Patent Application (PTO-152	

DETAILED ACTION

Priority

1. Acknowledgment is made of applicant's claim for foreign priority under 35 U.S.C. 119(a)-(d). Receipt is acknowledged of papers submitted under 35 U.S.C. 119(a)-(d), which papers have been placed of record in the file.

Information Disclosure Statement

2. The information disclosure statement (IDS) submitted on 08/05/02 was considered by the examiner.

Drawings

3. Figure 1 should be designated by a legend such as --Prior Art-- because only that which is old is illustrated. See MPEP § 608.02(g). A proposed drawing correction or corrected drawings are required in reply to the Office action to avoid abandonment of the application. The objection to the drawings will not be held in abeyance.

Specification

- 4. The preliminary amendment filed 06/17/02 does not introduce new matter into the disclosure.
- 5. The disclosure is objected to because of the following informalities: it is noted that the specific makes reference to several copending applications, see page 7, in this case, such reference must include the relationship (i.e., continuation, divisional, or continuation-in-part) of the applications, if applicable; also, the status of nonprovisional application(s) (whether patented

or abandoned) should also be included. If a parent application has become a patent, the expression "now Patent No. _____" should follow the filing date of the application. If a parent application has become abandoned, the expression "now abandoned" should follow the filing date of the parent application. Appropriate correction is required.

Claim Rejections - 35 USC § 102

6. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

With respect to claim 1:

- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- 7. Claims 1-21 are rejected under 35 U.S.C. 102(b) as being anticipated by Okada et al 5302470.

The present application is directed to a process for producing electricity in a fuel cell wherein the disclosed inventive concept comprises the specific reforming temperature.

Okada et al disclose a desulfurized raw fuel material 1 that is mixed with steam at an appropriate ratio in a mixer 3 and transferred to a steam reformer 4 where it is converted by stream reforming reaction to a fuel gas consisting of hydrogen. The fuel gas is further transferred to a fuel electrode 7 in a fuel cell unit 6 where it is partially consumed by electrochemical reaction with a supply of air 9 fed by a compressor 8 to an oxidant electrode 10 of the fuel cell unit 6 so that electricity is generated while water is released (COL 8, lines 20-37). **EXAMPLE 1** shows that the reforming reaction takes place at reaction temperatures of 450 °C (COL 11, line

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50-60). **EXAMPLES 1-7** show the use of town gas, naphta, LPG among others (See EXAMPLES 1-7). Thus, the fuels are higher carbon (C_{2+}) hydrocarbons. **TABLE 1** also shows methane content of 86.9 % by volume (See Table 1).

With respect to claims 2-3 and 16-17:

Okada et al shows methane content of 86.9 % by volume (See TABLE 1).

As to claims 4, 18-19:

Okada et al teach the steam reforming reaction at least at temperature of <u>about</u> 450 °C (COL 7, lines 11-16). It is also disclosed a temperature range of 350-400 °C (COL 7, line 5-7). With respect to claim 5:

EXAMPLE 7 shows adiabatic conditions (refer to TABLE 2) as well as quasi-adiabatic conditions (COL 15, line 25-27).

With respect to claims 6 and 20-21:

It is disclosed that the S/C ratio is at least 0.7, or 1.5 depending on the specific catalyst material (COL 4, lines 53-65/COL 7, lines 17-22/COL 9, lines 17-22). Thus, Okada et al directly teach the use of the S/C ratio within the claimed ratio.

As to claims 7-8 and 13:

It is disclosed that surplus steam remains unused during the steam reforming reaction (COL 3, lines 35-40). It is also disclosed that the fuel gas discharged from the fuel electrode 7 is transferred to a burner 11 in the steam reformer 4 where it is mixed therewith and burned for heating the steam reformer (COL 2, lines 42-47). It is also disclosed that the separated steam is transferred to the mixer 3 where it is mixed with the raw fuel material 1, then, fed to the steam

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reformer 4 for use in the steam reforming reaction (COL 2, lines 55-60). It is disclosed that a reactive gas after electrode reaction is recovered for reuse (COL 9, lines 57-64).

On the matter of claims 9-11:

Okada et al discloses the use of raw fuel material selected from methane, ethane, propane, butane, natural gas, naphta, kerosene, gas oil, LPG, town gas and their mixtures (COL 7, lines 32-35/ COL 1, lines 35-40/ ABSTRACT/ EXAMPLES 1-7).

As to claim 12:

Okada et al disclose that high temperature fuel cell e.g. a molten carbonate fuel cell or a solid oxide fuel cell are types of fuel cell that can be employed for this purposes (COL 7, lines 40-44). Thus, the specific temperature reaction is inherent to these fuel cell and their operations. With respect to claims 14-15:

Okada et al teach that the raw fuel material is mixed with steam and converted by the steam reforming reaction in a steam reformer to a fuel gas (COL 8, line 65 to COL 9, line 3). It is evident from <u>TABLES 3-6</u> that a complete conversion of the higher carbon (C_{2+}) hydrocarbon occurs as the composition of the fuel gas shows no content of the higher carbon (C_{2+}) hydrocarbon (See TABLES 3-6).

Thus, Okada et al anticipates the instant claims.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Raymond Alejandro whose telephone number is (703) 306-3326.

The examiner can normally be reached on Monday-Thursday (8:30 am - 7:00 pm).

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick J. Ryan can be reached on (703) 308-2383. The fax phone number for the organization where this application or proceeding is assigned is (703) 872-9310.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is (703) 308-0661.

Raymond Alejandro

Examiner

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